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## **IGCSE Chemistry 2012 Past Papers - CIE Notes**

Paper 02 - Structured Essays Question 1



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Syllabus References: A: 7.1, 7.7, 7.3, 7.4; B2 7.1, 7.2, 7.3 Part (a) (i) tested candidates' knowledge of experimental procedure for collecting and measuring the volume of gas generated during a chemical reaction. Candidates were required to draw a suitable

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## **N S C O U N C I L**

Mark Scheme of Cambridge IGCSE  
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question paper. for the guidance of teachers. 0620 CHEMISTRY. 0620/61 Paper 6 (Alternative to Practical), maximum raw mark 60. This mark scheme is published as an aid to teachers and candidates, to indicate the requirements of the examination.

**0620 s12 ms 61 - Past Papers | GCE**

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## **Guide**

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## **9701 s12 ms 11 - Papers**

GCE Ordinary Level. MARK SCHEME for the May/June 2012 question paper. for the guidance of teachers. 5070 CHEMISTRY. 5070/42 Paper 4 (Alternative to Practical), maximum raw mark 60. This mark scheme is published as an aid to teachers and candidates, to indicate the requirements of the

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examination. It shows the basis on which Examiners were instructed to award marks.

## **5070 s12 ms 42 - WordPress.com**

0620 || 0971. You can find all CIE Chemistry IGCSE (0620 and 0971) Paper 6 past papers and mark schemes below. For 0971 Cambridge says 'this syllabus

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is graded from 9 to 1 but is otherwise the same as Cambridge IGCSE Chemistry - 0620.

## **CIE Paper 6 IGCSE Chemistry Past Papers - PMT**

060 || 0971. You can find all CIE Chemistry IGCSE (0620 and 0971) Paper 3 past papers and mark schemes below.

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For 0971 Cambridge says 'this syllabus is graded from 9 to 1 but is otherwise the same as Cambridge IGCSE Chemistry - 0620.

## **CIE Paper 3 IGCSE Chemistry Past Papers - PMT**

18 January 2019 : October / November 2018 papers are updated. Feb / March



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and May / June 2019 papers will be updated after result announcements.  
15/08/2019 : O Level Past Papers Of May and June are updated. 12/01/2020 : O Level Chemistry 2019 October/November Past Papers are updated. Chemistry Yearly Past Papers

**O Level Chemistry 5090 Past Papers**

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## **2019 June & Nov | CAIE ...**

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shows the basis on which Examiners were instructed to award marks.

## **9701 s12 ms 42**

MARK SCHEME for the May/June 2012 question paper. for the guidance of teachers. 0620 CHEMISTRY. 0620/31 Paper 3 (Extended Theory), maximum raw mark 80. This mark scheme is

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published as an aid to teachers and candidates, to indicate the requirements of the examination.

## **0620 s12 ms 31 - Past Papers | GCE Guide**

This question was attempted by majority of the candidates and the performance was fair. In (a)(i)(ii) and (iii) most

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candidates correctly define standard electrode potential, state factors that affect standard electrode potential and gave uses of the values of standard potential.

## **Chemistry Paper 2 (Essay) ,May/June 2012**

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01 — Multiple Choice This paper tested  
Sections A and B of the syllabus in the  
Knowledge and Comprehension profile  
dimension. Performance on this paper

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continues to be steady and satisfactory. The marks ranged from 0 to 60. The mean score was 51.5 per cent and the standard deviation was 11.39.

### **May June 2012 Chemistry Paper 62 0620**

It also contains complete answers to the questions set on the Chemistry Paper 02

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in the May/June series of examinations between the years 2012 and 2016. In addition to the answer to each question, an appropriate explanation is given so you don't just get the right answer but, more importantly, you can see how it has been worked out.

## **Chemistry Worked Solutions for**



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unambiguous. The questions were well  
structured and within the scope of the  
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scheme was detailed, clear and fair to all  
candidates. Chemistry Paper 2 (Essay)  
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June : 2019-Oct-Nov

## **Past Papers Of Home/Cambridge International Examinations ...**

A is solution containing 6.22 g of an acid  
H<sub>2</sub>Y per dm<sup>3</sup>. B contains 3.90 g of NaOH

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per  $\text{dm}^3$  of solution. (a) Put A into the burette and titrate it against  $20.0 \text{ cm}^3$  or  $25.0 \text{ cm}^3$  portions of B using methyl orange as indicator. Repeat the titration to obtain consistent titres. Tabulate your burette readings and calculate the average volume of acid A used.

## **Chemistry Paper 1 (Practical)**

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