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Section 7.3 A Bonding in Metals. The valence electrons of metal atoms can be modeled as a sea of electrons. Metallic bonds consist of the attraction of the free-floating valence electrons for the positively charged metal ions. Metals are good conductors and malleable because of their mobile electrons.

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number of valence electrons in an atom of a representative element. 3.

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978-0-13252-576-3,
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everywhere chemical deposits are located. The ruby crystal shown above is extremely valuable, both because of its beauty and its utility in equipment such as lasers.

Ionic Crystal Structures | Chemistry for Non-Majors

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affiliates. All Rights Reserved: 83. Ionic and Metallic Bonding. BONDING AND INTERACTIONS, 7.1 lons. EssentialUnderstanding lons form when atoms gain or lose valence electrons, becoming electrically charged. Lesson Summary. Valence ElectronsValence electrons are the electrons in the outermost occupied

energy level and are involved in ion swer formation. For a representative element, the group number equals the ...

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ionic and metallicallic Chapter 7: Ionics wer Compounds and Metals 206 Chapter 7 • Ionic Compounds and Metals Section 7711 Figure 71 As carbon dioxide dissolves in ocean water, carbonate ions are produced Coral polyps capture these carbonate ions, producing crystals of calcium

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Chapter 7 Ionic and Metallic Bonding157 25. State the number of electrons lost or gained in forming each of these ions. Name the ions and tell whether it is an anion or a cation.

05 CTR ch07 7/9/04 3:27 PM Page 155 IONS 7

Chapter 7 Ionic and Metallic Bonding 169 Name Date Class C 23. The cation Fe3+ is

formed when. a. an atom of iron loses two electrons. b. an atom of zinc loses two electrons. c. an atom of iron loses three electrons. d. an atom of iron gains three electrons. C. True-False.

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Compounds. Ionic
Bonding. Anions and
cations are held
together by . opposite.
charges (+ and -) Ionic
compounds are called .
salts. Formula unit Simplest . ratio. of
elements in an ionic
compound.

Chapter 7 Ionic and Metallic Bonding
Chemistry Chapter
7-Ionic and Metallic
Bonding. Feedback. We all know how hard

Chemistry can be allic Unless, of course you are that one child who seems to know everything already. Even with those kids, practice is key. Chemistry is easy to forget. However, if you can answer these correctly, you must be sharp!

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positively charged metal ions. Answer

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the representative element, simply look at the group number; 3

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